

PROBLEM SET SOLUTIONS

CHAPTER 9, Levine, *Quantum Chemistry*, 5th Ed.

9.1 For the anharmonic oscillator with the Hamiltonian

$$H = -\frac{\hbar^2}{2m} \left\{ \frac{d^2}{dx^2} \right\} + k \frac{x^2}{2} + c x^3 + d x^4$$

evaluate E^1 for the first excited state, taking the unperturbed system as the harmonic oscillator.

HINT: Example p.248 (5th Ed.) shows how to calculate E^1 for the ground state of the harmonic oscillator. Use the same method, just change the wavefunction to that for the first excited state.

$$E_N^1 = H_{NN}^1 = \int (\psi_N^0)^* H^1 \psi_N^0 d\tau, N = 1 \text{ for first excited state}$$

$$H^1 = H - H^0$$

$$H^0 = -\frac{\hbar^2}{2m} \left\{ \frac{d^2}{dx^2} \right\} + k \frac{x^2}{2}$$

$$H^1 = c x^3 + d x^4$$

For the harmonic oscillator, $\alpha = 2\pi\nu m/\hbar = 4\pi^2\nu m/h$ &

$\nu = 0$ is the ground state: $\psi_0 = c_0 e^{-\alpha x^2/2}$, $c_0 = (\alpha/\pi)^{1/4}$

Example p. 248 shows that the first order correction to the ground state energy of the anharmonic oscillator is

$$E_0^1 = H_{00}^1 = 3d/(4\alpha^2) = 3dh^2/[64\pi^4\nu^2m^2]$$

For the harmonic oscillator

$\nu = 1$ is the first excited state: $\psi_1 = c_1 x e^{-\alpha x^{**2/2}}$, $c_1 = (4\alpha^3/\pi)^{1/4}$

The first order correction to the energy of the first excited state of the anharmonic oscillator is

$$\begin{aligned} E_1^1 &= H_{11}^1 = \int (\psi_1^0)^* H^1 \psi_1^0 d\tau \\ &= \int_{-\infty}^{\infty} (c_1 x e^{-\alpha x^{**2/2}})^2 (cx^3 + dx^4) dx \\ &= (c_1)^2 \int_{-\infty}^{\infty} x^2 e^{-\alpha x^{**2}} (cx^3 + dx^4) dx \\ &= c (c_1)^2 \int_{-\infty}^{\infty} x^5 e^{-\alpha x^{**2}} dx + d (c_1)^2 \int_{-\infty}^{\infty} x^6 e^{-\alpha x^{**2}} dx \end{aligned}$$

first term is even x odd, so integral = 0

$$\begin{aligned} E_1^1 &= 2d (c_1)^2 \int_0^{\infty} x^6 e^{-\alpha x^{**2}} dx \\ &= 2d (c_1)^2 (15/2^4)(\pi/\alpha^7)^{1/2} \end{aligned}$$

$$= 2d (4\alpha^3/\pi)^{1/2}(15/2^4)(\pi/\alpha^7)^{1/2}$$

$$= d15/(4\alpha^2)$$

$$= d15/[4(4\pi^2vm/h)^2]$$

$$= d15h^2/[64\pi^4v^2m^2]$$