

PROBLEM SET SOLUTIONS

Chapter 7, *Quantum Chemistry*, 5th Ed., Levine

7.6 Which of the following operators are Hermitian?

For a Hermitian operator, $\langle A \rangle = \langle A \rangle^*$, or

$\langle f | A | g \rangle = \langle g | A | f \rangle^*$. Assume f & g are well-behaved at limits of integration.

Integration by parts: $\int u v' = uv - \int v u'$

$$\begin{aligned} \text{(a)} \quad \langle f | \frac{d}{dx} | g \rangle &= \int f^* (dg/dx) d\tau = f^* g \Big| - \int g (df/dx)^* d\tau \\ &= - \langle g | \frac{d}{dx} | f \rangle^* \quad \text{NO} \end{aligned}$$

$$\begin{aligned} \text{(b)} \quad \langle f | i \frac{d}{dx} | g \rangle &= \int f^* (i dg/dx) d\tau = i f^* g \Big| + \int g (i df/dx)^* d\tau \\ &= \langle g | i \frac{d}{dx} | f \rangle^* \quad \text{YES} \end{aligned}$$

$$\begin{aligned} \text{(c)} \quad \langle f | 4 \frac{d^2}{dx^2} | g \rangle &= \int f^* (4 d^2g/dx^2) d\tau ; [u = f, v' = d^2g/dx^2, \\ & \qquad \qquad \qquad v = dg/dx, u' = df/dx] \\ &= 4 f^* dg/dx \Big| - 4 \int (dg/dx) (df/dx)^* d\tau ; [u = df^*/dx, v' = dg/dx, \\ & \qquad \qquad \qquad v = g, u' = d^2f^*/dx^2] \\ &= -4 (df^*/dx)g \Big| + 4 \int g d^2f^*/dx^2 d\tau \\ &= \langle g | 4 \frac{d^2}{dx^2} | f \rangle \quad \text{YES} \end{aligned}$$

$$\begin{aligned} \text{(d)} \quad \langle f | i \frac{d^2}{dx^2} | g \rangle &= \int f^* (i d^2g/dx^2) d\tau ; [u = f, v' = d^2g/dx^2, \\ & \qquad \qquad \qquad v = dg/dx, u' = df/dx] \\ &= i f^* dg/dx \Big| - i \int (dg/dx) (df/dx)^* d\tau ; [u = df^*/dx, v' = dg/dx, \\ & \qquad \qquad \qquad v = g, u' = d^2f^*/dx^2] \\ &= -i (df^*/dx)g \Big| + i \int g d^2f^*/dx^2 d\tau = - \int g (i d^2f/dx^2)^* d\tau \end{aligned}$$

$$= -\langle g | \frac{d^2}{dx^2} | f \rangle^* \quad \text{NO}$$

7.9 Which of the following operators meet all the requirements for a quantum mechanical operator that is to represent a physical quantity?

Operator must be linear & Hermitian

(a) $\text{SQRT} = ()^{1/2}$ NOT LINEAR

(b) d/dx LINEAR, NOT HERMITIAN

(c) d^2/dx^2 LINEAR & HERMITIAN

(d) $i d/dx$ LINEAR & HERMITIAN

7.17 For the hydrogenlike atom,

$$V = -Z (e')^2 (x^2 + y^2 + z^2)^{-1/2}$$

And the potential energy is an even function of the coordinates.

(a) What is the parity of ψ_{2s} ?

$$\psi_{2s} = 1/[4(2\pi)^{1/2}] (Z/a)^{3/2} (2 - Zr/a) e^{-Zr/(2a)}$$

$$\Pi(x) = -x, \Pi(y) = -y, \Pi(z) = -z,$$

$$\Pi(r) = r, \Pi(\theta) = \pi - \theta, \Pi(\phi) = \pi + \phi$$

$$\Pi \psi_{2s} = 1/[4(2\pi)^{1/2}] (Z/a)^{3/2} \Pi \{ (2 - Zr/a) e^{-Zr/(2a)} \}$$

$$= 1/[4(2\pi)^{1/2}] (Z/a)^{3/2} (2 - Zr/a) e^{-Zr/(2a)}$$

$$= \psi_{2s} \quad \text{EVEN}$$

(b) What is the parity of ψ_{2px} ?

$$\psi_{2px} = 1/[4(2\pi)^{1/2}] (Z/a)^{5/2} r e^{-Zr/(2a)} \sin \theta \cos \phi$$

$$\Pi \psi_{2px} = 1/[4(2\pi)^{1/2}] (Z/a)^{5/2} \Pi \{ r e^{-Zr/(2a)} \sin \theta \cos \phi \}$$

$$= 1/[4(2\pi)^{1/2}] (Z/a)^{5/2} r e^{-Zr/(2a)} \sin(\pi - \theta) \cos(\pi + \phi)$$

$$\sin(\pi - \theta) = \sin \pi \cos \theta - \cos \pi \sin \theta = 0 - (-1) \sin \theta = \sin \theta$$

$$\cos(\pi + \phi) = \cos \pi \cos \phi - \sin \pi \sin \phi = -\cos \phi - 0 = -\cos \phi$$

$$\Pi \psi_{2px} = 1/[4(2\pi)^{1/2}] (Z/a)^{5/2} r e^{-Zr/(2a)} \sin \theta (-\cos \phi)$$

$$= -\psi_{2px} \quad \text{ODD}$$

$$(c) \psi_{2s} + \psi_{2px} = 1/[4(2\pi)^{1/2}] (Z/a)^{3/2} e^{-Zr/(2a)}$$

$$x \{ 2 - Zr/a + rZ/a \sin \theta \cos \phi \}$$

$$H(\psi_{2s} + \psi_{2px}) = H\psi_{2s} + H\psi_{2px} = E_2\psi_{2s} + E_2\psi_{2px}$$

$$= E_2(\psi_{2s} + \psi_{2px}) \quad \text{Yes, eigenfunction}$$

$$\Pi(\psi_{2s} + \psi_{2px}) = \Pi\psi_{2s} + \Pi\psi_{2px} = \psi_{2s} - \psi_{2px}$$

neither even nor odd, no parity

We showed previously that when V is even, the wavefunctions of a system with non-degenerate energy levels must be of definite parity. Here, the $n=2$ level is degenerate, hence no definite parity.

7.26 For a hydrogen atom in a p state, the possible outcomes of a measurement of L_z are $-\hbar$, 0, and \hbar . For each of the following wavefunctions give the probabilities of each of these three results.

$$L_z \psi_{2pm} = m \hbar \psi_{2pm}; \text{ for a p state, } m = -1, 0, 1$$

Write ψ as a linear combination of eigenfunctions of L_z . The probability of getting a particular value when the property is measured is the square of the corresponding coefficient.

$$\text{Probability of measuring property } i = |c_i|^2$$

$$1 = \sum |c_i|^2$$

$$(a) \psi_{2pz} = \psi_{2p0} = c_1 \psi_{2p-1} + c_2 \psi_{2p0} + c_3 \psi_{2p1}$$

$$c_1 = c_3 = 0, c_2 = 1$$

Probability of measuring \hbar is square of coefficient of ψ_{2p1} : 0

Probability of measuring $-\hbar$ is square of coefficient of ψ_{2p-1} : 0

Probability of measuring 0 is square of coefficient of ψ_{2p0} : 1.

$$\text{Note: } c_1^2 = c_2^2 + c_3^2 = 1 = 0 + 1 + 0$$

$$(b) \psi_{2py} = -i/\sqrt{2} \psi_{2p1} + i/\sqrt{2} \psi_{2p-1}$$

Probability of measuring \hbar is square of coefficient of ψ_{2p1} :

$$|i/\sqrt{2}|^2 = (-i/\sqrt{2})(-i/\sqrt{2})^* = 1/2$$

Probability of measuring $-\hbar$ is square of coefficient of ψ_{2p-1} :

$$|i/\sqrt{2}|^2 = (i/\sqrt{2})(i/\sqrt{2})^* = 1/2$$

Probability of measuring 0 is square of coefficient of ψ_{2p0} : 0

$$\text{Note: } c_1^2 = c_2^2 + c_3^2 = 1 = 1/2 + 1/2 + 0$$

$$(c) \psi_{2p1} = c_1 \psi_{2p-1} + c_2 \psi_{2p0} + c_3 \psi_{2p1}$$

$$c_1 = 0 = c_2, c_3 = 1 \text{ \& } c_1^2 = c_2^2 + c_3^2 = 1$$

Probability of measuring \hbar is square of coefficient of ψ_{2p1} : 1

Probability of measuring $-\hbar$ is square of coefficient of $\psi_{2p-1} : 0$

Probability of measuring 0 is square of coefficient of $\psi_{2p0} : 0$.

7.27 (3rd Ed.; like example, p. 185, 5th Ed.) Consider a particle in a nonstationary state in a one-dimensional box of length L with infinite walls. Suppose at time t_0 its state function is the parabolic function

$$\psi(t_0) = N x (L - x) \quad 0 \leq x \leq L$$

where N is the normalization constant. If at time t_0 we were to make a measurement of the particle's energy, what would be the possible outcomes of the measurement & what would be the probability for each such outcome?

For a 1D particle in a box, $H = -\hbar^2/(2m) d^2/dx^2$;

$V = 0$ ($0 \leq x \leq L$), $V = \infty$ ($x < 0$, $x > L$)

The complete set of eigenfunctions of the H operator for a 1D particle in a box are the ψ_n

$$\psi_n = (2/L)^{1/2} \sin(n\pi x/L) \quad 0 \leq x \leq L$$

$$\psi_n = 0 \quad x < 0, x > L$$

Since $\psi(t_0)$ is an arbitrary function, we can expand it in terms of the eigenfunctions of H :

$$\psi(t_0) = \sum c_n \psi_n, \text{ where } c_n = \langle \psi(t_0) | \psi_n \rangle.$$

The probability of obtaining the eigenfunction E_n when making a measurement is $|c_n|^2$. Find c_n :

$$\begin{aligned} c_n &= \langle \psi(t_0) | \psi_n \rangle \\ &= \int_0^L \psi(t_0) \psi_n dx = \int_0^L N x (L - x) (2/L)^{1/2} \sin(n\pi x/L) dx \\ &= N (2/L)^{1/2} \left\{ L \int_0^L x \sin(n\pi x/L) dx - \int_0^L x^2 \sin(n\pi x/L) dx \right\} \\ &= N (2/L)^{1/2} [L/(n\pi)]^3 \left\{ -(n\pi)^2 \cos(n\pi) \right. \\ &\quad \left. + [(n\pi)^2 - 2] \cos(n\pi) + 2 \right\} \\ &= N (2/L)^{1/2} [L/(n\pi)]^3 2 (1 - \cos(n\pi)) \end{aligned}$$

If $n = 1, 3, 5, \dots$, $\cos(n\pi) = -1$. If $n = 2, 4, 6, \dots$, $\cos(n\pi) = 1$.

$$c_n = N 2^{3/2} L^{5/2} / (n\pi)^3 (1 - (-1)) = N 2^{5/2} L^{5/2} / (n\pi)^3,$$

$$n = 1, 3, 5, \dots,$$

$$c_n = N 2^{3/2} L^{5/2} / (n\pi)^3 (1 - 1) = 0, \quad n = 2, 4, 6, \dots,$$

Probability of measuring E_n is $|c_n|^2$.

$$|c_n|^2 = 0, \quad n = 2, 4, 6, \dots,$$

$\psi(t_0) = N x (L - x)$, $0 \leq x \leq L$, is odd function

$\psi_n = (2/L)^{1/2} \sin(n\pi x/L)$, $0 \leq x \leq L$, is even & so doesn't contribute to $\psi(t_0)$

$$|c_n|^2 = N^2 2^5 L^5 / (n\pi)^6, \quad n = 1, 3, 5, \dots,$$

To evaluate c_n need normalization constant N :

$$\langle \psi(t_0) | \psi(t_0) \rangle = \int_0^L N^2 x^2 (L - x)^2 dx$$

$$= \int_0^L N^2 x^2 (L^2 - Lx + x^2) dx$$

$$= N^2 (L^2 \int_0^L x^2 dx - 2L \int_0^L x^3 dx + \int_0^L x^4 dx)$$

$$= N^2 \{L^2 (x^3/3) \Big|_0^L - 2L (x^4/4) \Big|_0^L + (x^5/5) \Big|_0^L\}$$

$$= N^2 \{L^5/3 - 2L^5/4 + L^5/5\}$$

$$= N^2 L^5 \{1/3 - 2/4 + 1/5\} = N^2 L^5 / 30 = 1, \quad \text{if } N = \text{SQRT}(30/L^5)$$

$$|c_n|^2 = (30/L^5) 2^5 L^5 / (n\pi)^6, \quad n = 1, 3, 5, \dots,$$

$$= (30)(32)/(n\pi)^6$$

$$|c_1|^2 = (30)(32)/(\pi)^6 = 0.99855$$

$$|c_3|^2 = (30)(32)/(2\pi)^6 = 0.001370$$

$$|c_5|^2 = (30)(32)/(4\pi)^6 = 0.000064$$

Most of the contribution comes from ψ_1 because it closely resembles $\psi(t_0)$ --See Fig. 7.3, p. 186,

5th Ed.

